

## Simplicity in Water Analysis

# **Cover Page for Safety Data Sheet**

Thank you for choosing CHEMetrics, Inc. We appreciate your business. In order to best serve your needs for accurate and complete Safety Data, we offer the following information as supplemental to the attached SDS.

SDS No.: R6002

Version No.: 1.1

Product Name: Iron in Brine CHEMets® Refill

**Part Nos.**: R-6002

## **Product Descriptions:**

CHEMets Refills: Sealed glass ampoules, 7 mm OD, for visual colorimetric water analysis. Each CHEMet™ ampoule contains approximately 0.5 mL of liquid reagent sealed under vacuum. Refills contain 30 ampoules, test kits contain 1 refill.

# Addendum to Section 14 Transport Information:

Shipping container markings and labels for this product, as received, may vary from the contents of section 14 of the SDS for one or both of the following reasons:

- CHEMetrics has packaged this product as Dangerous Goods in Excepted Quantities according to IATA, US DOT, and IMDG regulations.
- CHEMetrics has packaged this product as part of a test kit or reagent set composed of various chemical reagents and elected to ship as UN 3316 Chemical Kit, Hazard Class 9, Packing Group II or III.

In case of reshipment, it is the responsibility of the shipper to determine appropriate labels and markings in accordance with applicable transportation regulations.

### Additional Information:

- "Print Date" = Revision Date (expressed as DD/MM/YYYY)
- Test kits and reagents sets may contain additional chemical reagents. See separate SDS(s).



### Iron in Brine CHEMets Refill

CHEMetrics, Inc.

Chemwatch: 9-84704 SDS No: R6002 Version No: 1.1

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

### Chemwatch Hazard Alert Code: 2

Issue Date: **04/11/2014**Print Date: **12/03/2015**Initial Date: **05/11/2014**S.GHS.USA.EN

### SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

#### **Product Identifier**

Product name	Iron in Brine CHEMets Refill
Synonyms	Part No.: R-6002
Proper shipping name	Not Applicable
Chemical formula	Not Applicable
Other means of identification	Not Available
CAS number	Not Applicable

### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Component of water analysis test kit K-6002
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### Details of the manufacturer/importer

Registered company name	CHEMetrics, Inc.
Address	4295 Catlett Road, Midland, VA. 22728 United States
Telephone	1-540-788-9026
Fax	1-540-788-4856
Website	www.chemetrics.com
Email	technical@chemetrics.com

### Emergency telephone number

Association / Organisation	ChemTel Inc.
Emergency telephon number	e s 1-800-255-3924
Other emergency telephon number	e +01-813-248-0585

### **SECTION 2 HAZARDS IDENTIFICATION**

### Classification of the substance or mixture

Acute Toxicity (Oral) Category 4, Acute Toxicity (Dermal) Category 4, Acute Toxicity (Inhalation) Category 4, Chronic Aquatic Hazard Category 3

### Label elements

GHS label elements

**GHS Classification** 



SIGNAL WORD WARNING

#### Hazard statement(s)

• •	
H302	Harmful if swallowed
H312	Harmful in contact with skin
H332	Harmful if inhaled
H412	Harmful to aquatic life with long lasting effects

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### Iron in Brine CHEMets Refill

Wear protective gloves/protective clothing/eye protection/face protection.

P101	If medical advice is needed, have product container or label at hand.
P102	Keep out of reach of children.
P103	Read label before use.
P271	Use only outdoors or in a well-ventilated area.
P261	Avoid breathing dust/fume/gas/mist/vapours/spray.
P270	Do not eat, drink or smoke when using this product.
P273	Avoid release to the environment.

### Precautionary statement(s) Response

P280

P301+P312	IF SWALLOWED: Call a POISON CENTER/doctor/physician/first aider/if you feel unwell.	
P302+P352	IF ON SKIN: Wash with plenty of water and soap	
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.	
P330	Rinse mouth.	
P362+P364	Take off contaminated clothing and wash it before reuse.	

### Precautionary statement(s) Storage

Not Applicable

### Precautionary statement(s) Disposal

P501

Dispose of contents/container to authorised chemical landfill or if organic to high temperature incineration

### SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

#### Substances

See section below for composition of Mixtures

#### Mixtures

CAS No	%[weight]	Name
67-56-1	<2	<u>methanol</u>
1762-95-4	29	ammonium thiocyanate
7732-18-5	69	water

### **SECTION 4 FIRST AID MEASURES**

### Description of first aid measures

escription of first aid me	
	If this product comes in contact with the eyes:
<b>5 644</b>	Wash out immediately with fresh running water.
Eye Contact	• Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
	<ul> <li>Seek medical attention without delay; if pain persists or recurs seek medical attention.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
	If skin contact occurs:
Skin Contact	<ul> <li>Immediately remove all contaminated clothing, including footwear.</li> <li>Flush skin and hair with running water (and soap if available).</li> </ul>
	Seek medical attention in event of irritation.
	If fumes or combustion products are inhaled remove from contaminated area.
	Lay patient down. Keep warm and rested.
Inhalation	<ul> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if</li> </ul>
	necessary.
	► Transport to hospital, or doctor.
	▶ IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY.
	▶ For advice, contact a Poisons Information Centre or a doctor.
	▶ Urgent hospital treatment is likely to be needed.
	In the mean time, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicated by the patient's condition.
	▶ If the services of a medical officer or medical doctor are readily available, the patient should be placed in his/her care and a copy of the MSDS should be
	provided. Further action will be the responsibility of the medical specialist.
Ingestion	If medical attention is not available on the worksite or surroundings send the patient to a hospital together with a copy of the MSDS.
	Where medical attention is not immediately available or where the patient is more than 15 minutes from a hospital or unless instructed
	otherwise:
	▶ INDUCE vomiting with fingers down the back of the throat, ONLY IF CONSCIOUS. Lean patient forward or place on left side (head-down position, if
	possible) to maintain open airway and prevent aspiration.

# Indication of any immediate medical attention and special treatment needed

NOTE: Wear a protective glove when inducing vomiting by mechanical means.

Treat symptomatically.

For acute and short term repeated exposures to methanol:

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- Toxicity results from accumulation of formaldehyde/formic acid.
- Clinical signs are usually limited to CNS, eyes and GI tract Severe metabolic acidosis may produce dyspnea and profound systemic effects which may become intractable. All symptomatic patients should have arterial pH measured. Evaluate airway, breathing and circulation.
- ▶ Stabilise obtunded patients by giving naloxone, glucose and thiamine.
- Decontaminate with Ipecac or lavage for patients presenting 2 hours post-ingestion. Charcoal does not absorb well; the usefulness of cathartic is not established.
- Forced diuresis is not effective; haemodialysis is recommended where peak methanol levels exceed 50 mg/dL (this correlates with serum bicarbonate levels below 18 mEq/L).
- ▶ Ethanol, maintained at levels between 100 and 150 mg/dL, inhibits formation of toxic metabolites and may be indicated when peak methanol levels exceed 20 mg/dL. An intravenous solution of ethanol in D5W is optimal.
- Folate, as leucovorin, may increase the oxidative removal of formic acid. 4-methylpyrazole may be an effective adjunct in the treatment. 8. Phenytoin may be preferable to diazepam for controlling seizure.

[Ellenhorn Barceloux: Medical Toxicology]

**BIOLOGICAL EXPOSURE INDEX - BEI** 

DeterminantIndexSampling TimeComment1. Methanol in urine15 mg/lEnd of shiftB, NS2. Formic acid in urine80 mg/gm creatinineBefore the shift at end of workweekB, NS

B: Background levels occur in specimens collected from subjects NOT exposed.

NS: Non-specific determinant - observed following exposure to other materials.

For thiocyanate poisonings haemodialysis is recommended as the treatment of choice. Phenobarbital protects poisoned animals against death. Thiocyanate ion is slowly excreted in the urine and is not decomposed to any appreciable degree to cyanide.

[GOSSELIN, SMITH & HODGE: Clinical Toxicology of Commercial Products 5th Ed]

### **SECTION 5 FIREFIGHTING MEASURES**

#### **Extinguishing media**

The product contains a substantial proportion of water, therefore there are no restrictions on the type of extinguishing media which may be used. Choice of extinguishing media should take into account surrounding areas.

Though the material is non-combustible, evaporation of water from the mixture, caused by the heat of nearby fire, may produce floating layers of combustible substances.

In such an event consider:

- foam.
- ▶ dry chemical powder.

### Special hazards arising from the substrate or mixture

Fire Incompatibility

None known.

### Advice for firefighters

### Fire Fighting

- Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves in the event of a fire.
- ▶ Prevent, by any means available, spillage from entering drains or water courses.
- ▶ Use fire fighting procedures suitable for surrounding area.
- ▶ DO NOT approach containers suspected to be hot

#### Fire/Explosion Hazard

- ▶ The material is not readily combustible under normal conditions.
- $\blacksquare \ \ \, \text{However, it will break down under fire conditions and the organic component may burn}$
- Not considered to be a significant fire risk.
- ▶ Heat may cause expansion or decomposition with violent rupture of containers
- ▶ Decomposes on heating and may produce toxic fumes of carbon monoxide (CO).

### **SECTION 6 ACCIDENTAL RELEASE MEASURES**

### Personal precautions, protective equipment and emergency procedures

### Minor Spills

- ▶ Clean up all spills immediately.
- Avoid breathing vapours and contact with skin and eyes.
- ▶ Control personal contact with the substance, by using protective equipment.
- ▶ Contain and absorb spill with sand, earth, inert material or vermiculite
- Wipe up.

### Major Spills

- Moderate hazard.
- Clear area of personnel and move upwind.
   Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves.
- Prevent, by any means available, spillage from entering drains or water course.

Personal Protective Equipment advice is contained in Section 8 of the MSDS.

### **SECTION 7 HANDLING AND STORAGE**

#### Precautions for safe handling

### Avoid all personal contact, including inhalation.

#### ▶ Wear protective clothing when risk of exposure occurs.

# ► Use in a well-ventilated area. Safe handling ► Prevent concentration in hollo

- Prevent concentration in hollows and sumps.
- ► DO NOT enter confined spaces until atmosphere has been checked.

Wear impact- and splass-resistant eyewear. Break the ampoule tip only when it is completely immersed in sample. Breaking the tip in air may cause the glass ampoule to shatter.

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Other information

For optimum analytical performance, store in the dark and at room temperature.

#### Conditions for safe storage, including any incompatibilities

#### Suitable container

Storage incompatibility

- Polyethylene or polypropylene container.
- ▶ Packing as recommended by manufacturer
- ▶ Check all containers are clearly labelled and free from leaks.
- Metal cyanides are readily oxidised and those of some heavy metals show thermal instability.
  - Metal cyanide and cyanates are often endothermic, Several members of this family of compounds, containing heavy metals tend to explosive instability, and most are capable of violent oxidation under certain circumstances.
  - Fusion of mixtures of metal cyanides with metal chlorates, perchlorates, nitrates or nitrites cause violent explosion.
  - Magnesium reacts with incandescence on heating with several metal cyanides; release of cyanogen by thermal decomposition may cause vigorous reaction with magnesium.
  - Addition of one solid component (even in residual amounts) to another molten component is extremely dangerous.

#### PACKAGE MATERIAL INCOMPATIBILITIES

Not Available

### **SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION**

### **Control parameters**

#### OCCUPATIONAL EXPOSURE LIMITS (OEL)

#### INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Levels (PELs) - Table Z1	methanol	Methyl alcohol	260 mg/m3 / 200 ppm	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	methanol	Methanol	200 ppm	250 ppm	Not Available	TLV® Basis: Headache; eye dam; dizziness; nausea; BEI
US NIOSH Recommended Exposure Limits (RELs)	methanol	Carbinol, Columbian spirits, Methanol, Pyroligneous spirit, Wood alcohol, Wood naphtha, Wood spirit	260 mg/m3 / 200 ppm	325 mg/m3 / 250 ppm	Not Available	[skin]

### EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
methanol	Methyl alcohol; (Methanol)	Not Available	Not Available	Not Available
ammonium thiocyanate	Ammonium thiocyanate	8.6 mg/m3	54 mg/m3	54 mg/m3

Ingredient	Original IDLH	Revised IDLH	
methanol	25,000 ppm	6,000 ppm	
ammonium thiocyanate	Not Available	Not Available	
water	Not Available	Not Available	

### Exposure controls

# Appropriate engineering controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly.

### Personal protection











# Eye and face protection

- ▶ Safety glasses with side shields
- Chemical goggles
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available.

### Skin protection

### See Hand protection below

- ▶ Wear chemical protective gloves, e.g. PVC.
- ▶ Wear safety footwear or safety gumboots, e.g. Rubber

### Hands/feet protection

The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

Suitability and durability of glove type is dependent on usage.

#### Body protection

#### See Other protection below

# Other protection

- Overalls.P.V.C. apron.
- Barrier cream.

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Skin cleansing cream.

Thermal hazards

Not Available

#### Recommended material(s)

### GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

generated selection:

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Material	СРІ
BUTYL	A
NEOPRENE	В
BUTYL/NEOPRENE	С
NAT+NEOPR+NITRILE	С
NATURAL RUBBER	С
NATURAL+NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
PE/EVAL/PE	С
PVA	С
PVC	С
PVDC/PE/PVDC	С
SARANEX-23	С
SARANEX-23 2-PLY	С
TEFLON	С
VITON	С
VITON/NEOPRENE	С

<sup>\*</sup> CPI - Chemwatch Performance Index

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

#### Respiratory protection

Type AX Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	Air-line*	AX-2	AX-PAPR-2 ^
up to 20 x ES	-	AX-3	-
20+ x ES	-	Air-line**	-

<sup>\* -</sup> Continuous-flow; \*\* - Continuous-flow or positive pressure demand

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

#### **SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES**

### Information on basic physical and chemical properties

	• •		
Appearance	Colorless to pale green		
Physical state	Liquid	Relative density (Water = 1)	1.05
Odour	Odourless	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	5.3	Decomposition temperature	Not Available
Melting point / freezing point (°C)	0	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	100	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

### **SECTION 10 STABILITY AND REACTIVITY**

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

<sup>\*</sup> Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

<sup>^ -</sup> Full-face

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Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

### **SECTION 11 TOXICOLOGICAL INFORMATION**

Information	۸n	tovicale	leaine	offocts
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nformation on toxicologic	cal effects		
Inhaled	The material is not thought to produce respirat aerosols, especially for prolonged periods, may	generated by the material during the course of normal handling, may be harmful.  ry irritation (as classified by EC Directives using animal models). Nevertheless inhalation of vapours, fumes or produce respiratory discomfort and occasionally, distress.  the central nervous system, optic nerves and retinae. Symptoms may be delayed, with headache, fatigue,	
Ingestion	Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.		
Skin Contact	prolonged dermal exposures. Open cuts, abraded or irritated skin should not	as classified by EC Directives using animal models). Temporary discomfort, however, may result from e exposed to this material e, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use	
Eye	Although the liquid is not thought to be an irrita by tearing or conjunctival redness (as with wine	t (as classified by EC Directives), direct contact with the eye may produce transient discomfort characterised burn).	
Chronic	Based on experience with animal studies, there which do not cause significant toxic effects to th Long-term exposure to methanol vapour, at cor	occur and may cause some concern following repeated or long-term occupational exposure.  s a possibility that exposure to the material may result in toxic effects to the development of the foetus, at levels mother.  entrations exceeding 3000 ppm, may produce cumulative effects characterised by gastrointestinal disturbances s, insomnia, trembling, unsteady gait, vertigo, conjunctivitis and clouded or double vision. Liver and/or kidney	
Iron in Brine CHEMets Refill	TOXICITY	IRRITATION	
Iron in Brine CHEMets	TOXICITY	IRRITATION	

Iron in Brine CHEMets Refill	TOXICITY	IRRITATION
Iron in Brine CHEMets Refill	TOXICITY	IRRITATION

# Iron in Brine CHEMets

No significant acute toxicological data identified in literature search.

Goitrogenic:.

Goitrogens are substances that suppress the function of the thyroid gland by interfering with iodine uptake, which can, as a result, cause an enlargement of the thyroid, i.e., a goitre

- Goitrogens include: ▶ Vitexin, a flavanoid, which inhibits thyroid peroxidase thus contributing to goiter.
- ▶ Ions such as thiocyanate and perchlorate which decrease iodide uptake by competitive inhibition; as a consequence of reduced thyroxine and triiodothyronine secretion by the gland, at low doses, this causes an increased release of thyrotropin (by reduced negative feedback), which then stimulates the gland.
- ▶ Lithium which inhibits thyroid hormone release.

### **METHANOL**

The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.

WATER

No significant acute toxicological data identified in literature search.

Acute Toxicity	✓	Carcinogenicity	0
Skin Irritation/Corrosion	0	Reproductivity	0
Serious Eye Damage/Irritation	0	STOT - Single Exposure	0
Respiratory or Skin sensitisation	0	STOT - Repeated Exposure	0
Mutagenicity	0	Aspiration Hazard	0

Legend:

✓ – Data required to make classification available

🗶 – Data available but does not fill the criteria for classification

O - Data Not Available to make classification

#### **CMR STATUS**

SKIN methanol

 $US-Hawaii\ Air\ Contaminant\ Limits-Skin\ Designation | US-Alaska\ Limits\ for\ Air\ Contaminants-Skin\ Designation | US-A$  $NIOSH\ Recommended\ Exposure\ Limits\ (RELs)\ -\ Skin|US\ -\ Washington\ Permissible\ exposure\ limits\ of\ air\ contaminants$  Version No: **1.1** Page **7** of **8** Issue Date: **04/11/2014**Print Date: **12/03/2015** 

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- Skin|US - Michigan Exposure Limits for Air Contaminants - Skin|US - Tennessee Occupational Exposure Limits Limits For Air Contaminants - Skin|US ACGIH Threshold Limit Values (TLV) - Skin|US - California Permissible
Exposure Limits for Chemical Contaminants - Skin|US - North Carolina Permissible Exposure Limits (PELs) for Air
Contaminants - Skin Designation [NLV]|US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air
Contaminants - Skin|US - Minnesota Permissible Exposure Limits (PELs) - Skin

#### **SECTION 12 ECOLOGICAL INFORMATION**

#### Toxicity

Harmful to aquatic organisms, may cause long-term adverse effects in the aquatic environment

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters

Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

For Ammonia:

Atmospheric Fate: Ammonia reacts rapidly with available acids (mainly sulfuric, nitric, and sometimes hydrochloric acid) to form the corresponding salts.

#### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
methanol	LOW	LOW
ammonium thiocyanate	LOW	LOW
water	LOW	LOW

#### Bioaccumulative potential

Ingredient	Bioaccumulation
methanol	LOW (BCF = 10)
ammonium thiocyanate	LOW (LogKOW = 0.5829)
water	LOW (LogKOW = -1.38)

#### Mobility in soil

Ingredient	Mobility
methanol	HIGH (KOC = 1)
ammonium thiocyanate	LOW (KOC = 4.5)
water	LOW (KOC = 14.3)

### **SECTION 13 DISPOSAL CONSIDERATIONS**

#### Waste treatment methods

Product / Packag	ing
dispo	sal

Dispose of according to federal, state, and local regulations.

#### **SECTION 14 TRANSPORT INFORMATION**

#### Labels Required

**Marine Pollutant** 

NC

Land transport (DOT): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

### Transport in bulk according to Annex II of MARPOL 73 / 78 and the IBC code

Source	Ingredient	Pollution Category
IMO MARPOL 73/78 (Annex II) - List of Noxious Liquid Substances Carried in Bulk	methanol	Υ

### **SECTION 15 REGULATORY INFORMATION**

#### Safety, health and environmental regulations / legislation specific for the substance or mixture

methanol(67-56-1) is found on the following regulatory lists "US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants", "US - Idaho - Limits for Air Contaminants", "US - California Proposition 65 - No Significant Risk Levels (NSRLs) for Carcinogens", "US - Hawaii Air Contaminant Limits", "US - California Premissible Exposure Limits for Chemical Contaminants", "US - California Proposition 65 - Reproductive Toxicity", "US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants", "US - California - Proposition 65 - Priority List for the Development of MADLs for Chemicals Causing Reproductive Toxicity", "US - Oregon Permissible Exposure Limits (Z-1)", "US - Michigan Exposure Limits for Air Contaminants", "US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values", "US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)", "US NIOSH Recommended Exposure Limits (RELs)", "US - Alaska Limits for Air Contaminants", "US - Washington Permissible exposure limits of air contaminants", "US Priority List for the Development of Proposition 65 Safe Harbor Levels - No Significant Risk Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for Chemicals Causing Reproductive Toxicity", "US Spacecraft Maximum Allowable Concentrations (SMACs) for Airborne Contaminants", "US Minnesota Permissible Exposure Limits (PELs)", "US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)", "US Toxic Substances

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	Control Act (TSCA) - Chemical Substance Inventory", "US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants", "US OSHA Permissible Exposure Levels (PELs) - Table Z1"
ammonium thiocyanate(1762-95-4) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"
water(7732-18-5) is found on the following regulatory lists	"US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"

### **SECTION 16 OTHER INFORMATION**

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net/references

The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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